

# MCAT Prep Test: Inorganic Chemistry Flash Cards Cram Now MCAT Exam Review Study

Are you ready to ace your MCAT exam? Our Inorganic Chemistry Flash Cards are the perfect way to cram and review for your big day. With over 200 cards covering all the essential topics, you'll be able to quickly and easily learn all the inorganic chemistry you need to know.



## MCAT Prep Test INORGANIC CHEMISTRY Flash Cards-- CRAM NOW!--MCAT Exam Review Book & Study Guide (Cram Now! MCAT Study Guide 2)

★★★★☆ 4.2 out of 5

Language : English  
File size : 28100 KB  
Text-to-Speech : Enabled  
Screen Reader : Supported  
Enhanced typesetting: Enabled  
Print length : 1483 pages



Each card includes a concise definition of a key term, concept, or reaction, as well as a helpful image or diagram. The cards are organized by topic, so you can easily find the information you need. And because they're small and portable, you can study anywhere, anytime.

Why use our Inorganic Chemistry Flash Cards?

\* They're the perfect way to cram for your MCAT exam. \* They cover all the essential topics in inorganic chemistry. \* They're concise and easy to

understand. \* They're organized by topic, so you can easily find the information you need. \* They're small and portable, so you can study anywhere, anytime.

Don't wait until the last minute to start studying for your MCAT exam. Free Download your Inorganic Chemistry Flash Cards today and start cramming now!

### **Here's a sample of the cards you'll find in our Inorganic Chemistry Flash Cards:**

\* **Term:** Coordination complex \* **Definition:** A coordination complex is a molecule that contains a metal ion surrounded by ligands. \* **Image:** [Image of a coordination complex] \* **Term:** Ligand \* **Definition:** A ligand is a molecule, ion, or atom that donates a pair of electrons to a metal ion to form a coordination complex. \* **Image:** [Image of a ligand] \* **Term:** Oxidation state \* **Definition:** The oxidation state of an atom is the hypothetical charge that the atom would have if all of its bonds were ionic. \* **Image:** [Image of an oxidation state diagram]

... and many more!

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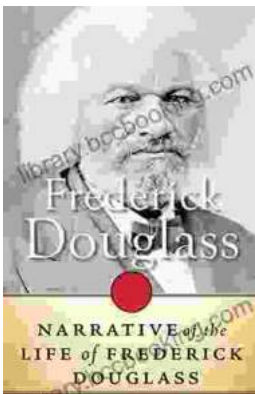
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